Total No. of printed pages = 6 BP 811 ET 23/08/22 Roll No. of candidate BINA CHOWDHURY CENTRAL LIBRAR Apere Halki-wapara, 2022 B.Pharm. 4th Year 8th Semester End-Term Examination ADVANCED INSTRUMENTATION TECHNIQUES (Theory) (New Regulation - w.e.f. 2017-18) Full Marks - 75 Time - Three hours The figures in the margin indicate full marks for the questions. Group - A Answer the following (MCQ): $(20 \times 1 = 20)$ (i) Chromatography is a physical method that is used to separate and analyse (a) Simple mixtures (b) Complex mixtures (c) Viscous mixtures -(d) Metals Beer's law states that the intensity of light decreases with respect to (ii)(a) Concentration (b) Distance (c) Composition (d) Volume (iii) Which of the following is not a type of Spectroscopy?

(a) Gamma-ray

(c) Nuclear Magnetic Resonance (NMR)

(b) X-ray

(d) Sound

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(iv)	Which of the following cannot be utilized as an adsorbent in column adsorption chromatography?					
	(a)	Magnesium oxide				
	(b)	Silica gel				
	(c)	Activated alumina				
	(d)	Potassium permanganate				
(v)	Wh	nich ratio does a mass detector estimate?				
	(a)	z/m				
	(b)	e/m				
	(c)	m/e				
	(d)	m/z				
(vi)	Wh	What is the purpose of mass spectroscopy?				
	(a)	Determination of molecule weight				
	(b)	Elucidating the chemical structures of molecules				
	(c)	(a) and (b)				
4	(d)	None of the above	ä			
(vii)	Wł	hich of the following solvent is not used in NMR?				
	(a)	D_2O				
	(b)	CHCl ₃	84			
	(c)	CCl ₄				
¥	(d)	CDCl ₃				
(viii) Ch	nemical shift has the units of				
	(a)	Parts per million				
	(b)	Parts per billion	-			
	(c)	Tesla				
	(d)) No units				
(ix)	TGA used to measure ———.					
	(a)	Change in mass-loss of weight				
	(b)	Change in temperature reaction				
	(c)	Pressure				
	(d)	(b) and (c)				

- (x) HPLC is an abbreviation for
 - (a) High-Profit Liquid Chromatography
 - (b) High-Pressure Liquid Chromatography
 - (c) Higher-Performance Low Chromatography
 - (d) Higher-Profit Low Chromatography
- (xi) In reverse phase chromatography, the stationary phase is made
 - (a) Non-polar
 - (b) Polar
 - (c) Either non-polar or polar
 - (d) None of these
- (xii) RP-HPLC method contains
 - (a) Stationary phase is polar, and mobile phase is non-Polar
 - (b) Stationary phase is non-polar, and mobile phase is polar
 - (c) Stationary phase is non-polar, and mobile, Phase is non-polar
 - (d) Stationary phase is polar, and mobile phase is polar
- (xiii) Which of the statements is correct?
 - (a) Gas chromatography is used to analyse gases
 - (b) Gas chromatography, is used to analyse solids
 - (c) Gas chromatography is used to analyse gases, solutions and solids
 - (d) All of the above
- (xiv) Theoretical plates are used to
 - (a) Estimate the efficiency of a column
 - (b) Determine the thickness of the stationary phase
 - (c) Measure the distribution of the analyte between mobile and stationary phases
 - (d) None of the above
- (xv) Mass spectrometer separates ions on the basis of which of the following?
 - (a) Mass
 - (b) Charge
 - (c) Molecular weight
 - (d) Mass to charge ratio

(xvi)	ICE	I guideline is used for Analytical method validation
	(a)	Q2
	(b)	Q7
	(c)	Q8
	(d)	Q1A-Q1F
(xvii)	The	e reference standard used, in NMR spectroscopy is
	(a)	Tetramethyl silane
	(b)	Benzene
	(c)	Ethanol
	(d)	C18 silica gel
(xviii)The	e highest m/z peak in the mass spectrum is called a
	(a)	Base peak
	(b)	Fragment peak
	(c)	Isotopic peak
	(d)	Parent peak
(xix)	IR	spectroscopy generally used to determine
	(a)	Molecular structure
	(b)	Functional group
	(c)	Number of protons
	(d)	All of the above
(xx)	TO	GA curve is a plot
	(a)	Weight vs. temperature
	(b)	Weight vs. volume of titrant
	(c)	Weight vs. current
	(d)	Weight loss vs. temperature

(Short Answer questions)

2. Answer any seven questions.

 $(7 \times 5 = 35)$

- (a) Write down the principles and application of NMR spectroscopy.
- (b) Write a short note on Radio-immune assay (RIA).
- (c) How will you differentiate the following pair of compounds by NMR spectroscopy?

- (d) What is the difference between DSC and TGA?
- (e) Write a short note on gas chromatography (GC) validation.
- (f) Write briefly about hyphenated chromatographic techniques used in drug analysis.
- (g) Write a short note on ionization techniques used in mass spectrometry.
- (h) Discuss the principle of X-ray crystallography and its application.
- (i) What do you mean by DTA and mention its applications?

Group - C

(Long Answer questions)

3. Answer any two of the following.

 $(2 \times 10 = 20)$

(a) (i) What is ICH guideline?

(3+7)

(ii) Briefly explain how can HPLC be calibrated and validated as per ICH and USFDA guidelines?