

Total No. of printed pages = 3

BINA CHOWDHURY CENTRAL LIBRARY

(IIMT & IIPS)
Aare Hatki, Wapara,
Wahaz-721017

ME 181501

Roll No. of candidate

--	--	--	--	--	--	--	--	--	--

17/12/22 2021

B.Tech. 5th Semester End-Term Examination

ME

APPLIED THERMODYNAMICS - I

(New Regulation & New Syllabus)

Full Marks – 70

Time – Three hours

The figures in the margin indicate full marks
for the questions.

1. Answer any four questions from the followings: $(4 \times 2\frac{1}{2} = 10)$
- Define four thermodynamic properties of steam with their SI units for solving engineering problem.
 - Differentiate between adiabatic process and isothermal process.
 - Draw the P-V and T-S diagram of a basic steam power plant cycle.
 - Draw T-S diagram of two improved methods of basic steam power plant cycle.
 - State two laws of thermodynamics that are applied in a steam power plant.
 - Construct a rough sketch of Mollier diagram and show important lines in it.
2. Answer any four questions from the followings: $(4 \times 5 = 20)$
- Draw a suitable sketch of Cochran boiler and label the various components.
 - Express the various methods to measure the evaporating capacity of a steam generator.
 - Describe in brief the influence of convergent and divergent part of a nozzle on steam velocity.

[Turn over

- (d) Deduce the expression of critical pressure ratio in terms of index of expansion for maximum discharge condition of steam flow through a nozzle.
- (e) Classify steam turbine on the basis of (i) action of steam and (ii) direction of steam flow.
- (f) Illustrate the working of steam condenser with suitable diagram.

3. Answer any four questions from the followings: (4 × 5 = 20)

- (a) A steam generator evaporates 18200 kg/hr of steam at 14 bar and a quality of 0.98 from feed water at 102°C when coal is burned at the rate of 2050 kg /hr. If the efficiency of the boiler is 63% then evaluate the caloric value of coal in kJ/kg used for steam generation.
- (b) Steam is expanded frictionless adiabatically in a steam nozzle from 5 bar and 400°C to 1 bar. If the initial velocity of steam entering the nozzle is 90 m/sec then evaluate the exit velocity of steam.
- (c) In a steam turbine, steam is supplied at 12 bar and dry-saturated. The condenser pressure is 0.8 bar. Determine the efficiency of the steam power cycle.
- (d) Estimate the flow rate of cooling water in kg per hour in a condenser which takes 12,500 kg of steam per hour. The steam enters at 0.08 bar with dryness fraction 0.94. The condensate leaves at 38°C. The inlet temperature and outlet temperature of water flowing through the condenser are 12°C and 22°C respectively.
- (e) Evaluate the value of blade efficiency of the impulse turbine from following data available:

Mean diameter of blade=1.2 m

Turbine speed=3000 RPM

Nozzle angle=20°

Inlet blade angle=25°

Outlet blade angle=18°

Blade velocity coefficient=0.92

- (f) Determine the height of the chimney of a boiler based on the following data:

Equivalent draught produced=18 mm of water.

Flue gas temperature=278°C

Ambient temperature=25°C

Minimum amount of air per kg of fuel=17230 gm

PUNA CHRONICLE CENTRAL LIBRARY
 10/11/2023
 10/11/2023
 10/11/2023

4. Answer any four questions from the followings: (4 × 2½ = 10)

- (a) Distinguish between fire-tube boiler and water tube boiler.
- (b) Differentiate between boiler mountings and boiler accessories.
- (c) Distinguish between the working of impulse turbine and reaction turbine.

- (d) Distinguish between Jet condenser and surface condenser.
- (e) Compare the relative advantages between different methods of steam turbine governing.
- (f) Compare the working performance between Back pressure turbine and pass-out turbine.

5. Answer any TWO questions from the followings: (2 × 5 = 10)

- (a) Deduce that unavailable energy is the product of lowest temperature of heat rejection and the change of entropy during the process.
- (b) Illustrate Keenan function and Gibb's function with their mathematical expressions.
- (c) Explain the terms Irreversibility and Effectiveness.
- (d) Differentiate between available energy and unavailable energy

BINA CHOWDHURY CENTRAL LIBRARY
ME 181501
2022